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Special Focus on Two-Dimensional Materials and Device Applications

# Properties and photodetector applications of two-dimensional black arsenic phosphorus and black phosphorus

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Abstract Two-dimensional (2D) black arsenic phosphorus (b-AsP), as an alloy of black phosphorus (b-P) with arsenic, has attracted great attention because of its outstanding electronic and optical properties, including high carrier mobility, tunable bandgap and in-plane anisotropy. B-AsP has a smaller bandgap (0.15–0.3 eV) than the b-P bandgap (0.3–2.0 eV), and thus can be used for mid-infrared photodetectors. In addition, both of them can form various van der Waals (vdW) heterojunctions with other 2D materials to realize novel functional optoelectronic devices. Here, we compare the basic characteristics of b-AsP and b-P, including crystal structure, optical properties, band structure, electrical properties and stability, and we summarize the update progress of b-AsP in photo detection, including representatives of phototransistor and photodiode devices. In the last part, the future research directions are discussed.

 ${\bf Keywords} \quad {\rm black\ arsenic\ phosphorus,\ crystal\ structure,\ optical\ property,\ electrical\ property,\ photodetector$ 

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# 1 Introduction

Black phosphorus (b-P) was synthesized by phase transition of white phosphorus under high pressure in 1914 [1], and it was reported that the bulk b-P was a kind of semiconductor in 1986, but its quality was too difficult to control and it hardly attracted people's interest [2]. Mechanical exfoliation and transfer technology has developed rapidly since graphene was discovered [3–6], and two-dimensional (2D) b-P has received widespread attention [7]. In 2014, several groups revealed the properties of monolayer or few-layer b-P through theoretical calculations and experiments, which shows outstanding properties, such as tunable bandgap [8], in-plane anisotropy [9] and high carrier mobility [10].

Interestingly, by introducing arsenic into b-P using a high pressure process, a new material named black arsenic phosphorus (b-AsP) can be synthesized [11], and this bulk material exhibits superconducting characteristics at a low temperature of about 10 K. In 2015, layered b-AsP was first reported as an infrared semiconductor with tunable composition and bandgap (0.15–0.3 eV) [12], so it is suitable for near- and middle-infrared photo detection. At the same time, layered b-AsP can form various van der Waals (vdW) heterojunctions with other 2D materials, thereby providing a method for fabricating a new type of high-performance photodetectors [13–16].

Compared with the widely studied b-P, there are fewer studies on b-AsP. In this forward-looking review, we adopt a strategy to compare b-P and b-AsP from the basic properties to applications in the photo detection field, aiming to provide some promising directions for the future research on b-AsP. We

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Figure 1 (Color online) Crystal structures of b-P and b-AsP. (a-i) Perspective view of b-P crystal structure. The spacing of interlayer is 0.53 nm. (a-ii) Top view of monolayer b-P, where x and y correspond to the directions of armchair and zigzag, respectively. (b-i) Perspective view of b-AsP crystal lattice. (b-ii) Crystal lattice of the monolayer b-AsP in top and side views. (b-ii) is reproduced with permission from [21].

first introduce the basic properties of b-P and b-AsP, including crystal structure, optical properties, band structure, electrical properties and stability. Subsequently, we summarize the update of b-P and b-AsP for photo detection, such as phototransistors and photodiodes. Finally, the future research directions are discussed.

# 2 **Properties**

#### 2.1 Crystal structure

To understand the crystal of b-AsP, the structure of b-P should be considered. In a monolayer b-P, there are two atomic layers including two kinds of P-P bonds, which are divided into the short bond and the long bond by length. As for the short bond in the crystal, which is connected by the two nearest phosphorus atoms in the same atomic layer with a bond length of 0.2224 nm. While the long bond consists of two nearest phosphorus atoms from top and bottom atomic layer, showing a bond length of 0.2244 nm (Figure 1(a-i)). As shown in Figure 1(a-ii), from the top view, the lattice of b-P shows an irregular hexagonal structure with two kinds of bond angles. The larger bond angle is consisted of two short bonds, showing a value of  $102.1^{\circ}$ , and the smaller angle is consisted of a short bond and a long bond, showing a value of  $96.3^{\circ}$  [17, 18]. In addition, the puckered structure of monolayer b-P reduces the symmetry of the parent isotropic structure, and only one two-fold rotation and one mirror plane are contained in the lattice structure [19, 20] (Figure 1(a-ii)).

The atomic structure of b-AsP is similar to that of b-P, shown in Figure 1(b-i). This crystal also has a puckered structure with irregular hexagonal structure. From the top view, the two nearest phosphorus or arsenic atoms along the armchair direction show a crystal constant of 4.60 Å, while the two nearest phosphorus or arsenic atoms along the zigzag direction show a crystal constant of 3.51 Å (Figure 1(bii)) [21]. In addition to theoretical calculations, experimental data are more credible. The crystal structure of b-AsP can be obtained by characterizing the high resolution transmission electron microscope (TEM). From the high resolution TEM image of the b-AsP flake, the orthorhombic crystal structure is shown and the crystal constant along different directions can be measured. The crystal constant along armchair



Figure 2 (Color online) Band structures of b-P and b-AsP. (a-i) Band structures of one-layer, two-layers, three-layers and bulk phosphorene calculated using density functional theory. (a-ii) The relationship between bandgap and the layer number in theory and experiment. (b-i) b-AsP's orbital-resolved band structure obtained from first principles calculations coupled with the function formalism of non-equilibrium green. (b-ii) Component of arsenic dependent bandgaps of thick b-As<sub>x</sub>P<sub>1-x</sub> flakes (> 30 nm). (b-iii) Bandgaps of b-As<sub>x</sub>P<sub>1-x</sub> determined by the arsenic component and number of layers, calculated by HSE06 method. (a-i) is reproduced with permission from [30], (a-ii) from [8], (b-i) from [21], (b-ii) from [12], and (b-iii) from [31].

direction is 3.3 Å and along zigzag direction is 4.7 Å [22], which is agree with the calculation results [21]. The crystal constant of b-P along armchair direction is about 4.4 Å, which is smaller than that of b-AsP, indicating that the increase of arsenic atoms leads to an increase of the lattice constant. Therefore, b-AsP has the similar crystal structure as b-P, but a larger lattice constant than b-P [22]. The puckered honeycomb configuration of b-AsP results in its highly anisotropy electronic [23], thermal [24–26] and optical properties [12, 27–29].

#### 2.2 Band structure

Figure 2 [8, 12, 21, 30, 31] shows the band structures of b-P and b-AsP. Monolayer and bulk b-P show all direct bandgaps [8, 32]. With the increase number of layers, the bandgap of b-P reveals a redshift which decreases from 2.0 eV (monolayer) to 0.3 eV (bulk) (Figure 2(a-i)) [30]. The bandgap as a function of the thickness is calculated by the GW method, as shown in Figure 2(a-ii). Experimental and theoretical bandgaps of all materials exhibit some differences, which are ascribed to different conditions [8, 33].

For the band structure of monolayer b-AsP, several theoretical calculations show a direct bandgap at around 0.92 eV [21, 23, 34]. Sun et al. [23] calculated the bandgap of monolayer b-AsP using first principles calculations method, which ranges from 0.92 to 0.95 eV owing to the increase of the component of phosphorus. Zhou et al. [21] reported similar results from first principles calculations coupled with the function formalism of non-equilibrium green (Figure 2(b-i)). In thick b-As<sub>x</sub>P<sub>1-x</sub> materials (>30 nm), the bandgap is decreased with increasing arsenic content (Figure 2(b-ii)) [12]. The bandgap of b-As<sub>x</sub>P<sub>1-x</sub> is decreased sharply when the arsenic content increases from x = 0 to 0.25 and decreases more slowly when the arsenic content increases further to x = 0.4 and finally to x = 0.83. At the same time, the bandgaps of b-AsP with the same composition are also different. Each date point in Figure 2(b-ii) corresponds to the measurement result from the same b-AsP flakes using different polarization angles or from b-AsP flakes with the same composition but different thicknesses, resulting in different of bandgap values. Wu et al. [14] reported a clear p-type band structure of multi-layer b-AsP (10.8 nm) with a bandgap of 0.25 eV and a work function of 4.4 eV. However, some groups calculated smaller or larger b-As P bandgaps than 0.92 eV [26,28]. The relationship between material thickness and bandgaps is shown in Figure 2(b). As the thickness increases, the bandgaps of these two materials will decrease obviously (Figure 2(b-iii)). In general, the bandgap monolayer, few-layer, and bulk b-AsP are smaller than that of b-P [31].



Figure 3 (Color online) Optical properties of b-P and b-AsP. (a-i) An atomic force microscopy image of a thin b-P flake, shows a thickness about 7.75 nm. Inset: optical image of this b-P flake. (a-ii) Raman spectrum of b-P using polarized laser excitation along different directions. (b-i) Raman spectra of b-AsP with different contents of arsenic. (b-ii) Polarized infrared extinction spectra of the b-As<sub>0.83</sub>P<sub>0.17</sub>. Inset: optical image of the characterized flake. (a-i) and (a-ii) are reproduced with permission from [35], (b-i) from [12], and (b-ii) from [29].

# 2.3 Optical properties

Three peaks at 365, 440, and 470 cm<sup>-1</sup> correspond to the  $A_g^1$ ,  $B_{2g}$ , and  $A_g^2$  vibration modes of b-P, respectively [35–41]. As shown in Figure 3(a-i), an atomic force microscopy image of a thin b-P flake shows a thickness of 7.75 nm [35]. An optical image of this b-P is shown in the inset of Figure 3(a-i). The Raman spectra of this flake obtained at different polarizations is shown in Figure 3(a-ii). The intensity of the  $A_{2g}$  mode decreases obviously with increasing polarization of the excitation laser from  $0^{\circ}(x)$  to  $90^{\circ}(y)$ , because the main atomic vibration of the  $A_{2g}$  mode is in the *x* direction (armchair). So that the crystal orientation of monolayer b-P can be identified by polarization-resolved Raman spectroscopy [35]. When the b-P flake is illuminated by a polarized incident infrared light from the *z* direction, and the incident light is polarized along six directions, which ranges from 0° to 330° with a step of 30°, all polarized directions of the infrared spectra show obvious increase at 2400 cm<sup>-1</sup>, corresponding to a bandgap of 0.3 eV [19].

Three major regimes are divided in the Raman spectra of b-AsP according to frequency. The low-frequency, medium-frequency and high-frequency regimes correspond to 200–300 cm<sup>-1</sup>, 300–380 cm<sup>-1</sup> and 380–500 cm<sup>-1</sup>, respectively. For b-As<sub>0.83</sub>P<sub>0.17</sub>, three peaks at 256, 233, and 224 cm<sup>-1</sup>are detected in the low-frequency regime, which are assigned to the  $A_g^2$ ,  $B_{2g}$  and  $A_g^1$  modes, respectively (Figure 3(b-i)) [12,28]. As the increase of phosphorus content, a gradual redshift of peaks is discovered. In comparison with b-P, b-AsP exhibits more Raman peaks, which is agreement with the existence of heteroatomic P-As bond (Figure 3(b-i)) [12]. As shown in Figure 3(b-ii) inset, a 188 nm-thick b-As<sub>0.83</sub>P<sub>0.17</sub> flake is fabricated using a mechanical exfoliation method. The polarization-resolved extinction spectra of this flake is measured using an infrared light, and shows an obvious increase at around 1000 cm<sup>-1</sup>, corresponding to the bandgap of 0.124 eV [29]. The family of b-As<sub>x</sub>P<sub>1-x</sub> with different arsenic content (x = 0, 0.25, 0.4, and 0.83) is studied using polarization-resolved infrared absorption systematically [12]. As the increase of x in b-As<sub>x</sub>P<sub>1-x</sub>, the absorption edges of b-AsP shift to shorter wavelengths clearly, indicating that the bandgap of b-As<sub>x</sub>P<sub>1-x</sub> is decreased with increasing the amount of arsenic.





Figure 4 (Color online) Electronic properties of b-P and b-AsP. (a-i) Schematic of b-P device structure with eight electrodes along with different directions. (a-ii) The  $I_{ds}$  and the transconductance as a function of angle. (b-i) Schematic of a b-AsP based field-effect transistor. (b-ii) Transfer curve of a thin b-As<sub>0.83</sub>P<sub>0.17</sub> flake in semilog scale and linear scale (inset). (a-i) and (a-ii) are reproduced with permission from [9], (b-ii) from [12].

#### 2.4 Electronic properties

In 2014, Liu et al. [9] first reported a field effect transistor based on few-layer b-P. In this device, a hole mobility of 286 cm<sup>2</sup>·V<sup>-1</sup>· s<sup>-1</sup> and an on/off ratio of  $10^4$  are achieved. The electrical transport of this device along different directions is different, showing an anisotropic property (Figures 4(a-i) and (a-ii)). Almost at the same time, a 10 nm-thick b-P field-effect transistor was fabricated by Li et al. [10], showing an high hole mobility about 1000 cm<sup>2</sup>·V<sup>-1</sup>· s<sup>-1</sup> and a high on/off ratio in the order of  $10^5$ . These studies initiated the research on b-P based nanoelectronic applications [33, 40–58]. To further improve the performance of b-P based field-effect transistors, devices with different thicknesses, contact electrodes, dielectric properties, and passivation layers have been fabricated (see details in Table 1 [9, 10, 12, 13, 19, 29, 42, 44, 46, 50, 52, 53, 56, 58-74]. Table 1 shows that transistors with ultrahigh hole mobility are mainly based on Cr/Au contact [10, 66], and the mobility is further improved by hBN passivation [68, 70]. For b-P based field-effect transistors, intrinsic ambipolar transfer characteristics have been widely observed owing to electrostatic modulation under an external electrical field, while the ambipolar behavior of b-P is always asymmetrical because of much lower electron concentration and mobility. The most direct way to improve the electron mobility of b-P is by n-type doping with  $Cs_2CO_3$ , Al, Cu, and other dopants [52–58, 65, 66, 68, 70, 72, 73]. By contact-metal engineering, n-type field-effect transistors can also be realized. For example, black phosphorus with Al contact and a flake thickness of 13 nm exhibits unipolar to ambipolar transition with an electron mobility of up to 950 cm<sup>2</sup>·V<sup>-1</sup>·s<sup>-1</sup> [71].

Field-effect transistors based on b-AsP typically have on/off ratios lower than  $10^4$  owing to their small bandgaps. As shown in Figure 4(b-i), a field-effect transistor is fabricated using a thin b-As<sub>0.83</sub>P<sub>0.17</sub> flake. The transfer curve of this device is shown in Figure 4(b-ii) [12]. The device has an ambipolar transport behavior with an on/off ratio about  $1.9 \times 10^3$ . By increasing the thickness from 11 to 21 nm, the on/off current ratio of b-AsP is decreased rapidly from  $10^3$  to 1.2 [16]. Although theoretical calculation shows that the mobility of b-AsP is higher than  $10000 \text{ cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1}$  [69], the experimental mobility is in the range of 80–310 cm<sup>2</sup> · V<sup>-1</sup> · s<sup>-1</sup> [13, 29]. As shown in Table 1, the mobility of b-AsP is much lower than that of b-P. Therefore, improvement in electrical performance of b-AsP is highly needed.

#### 2.5 Material stability

For the production of new devices from emerging materials, the long-term stability of devices needs to be first demonstrated [75–77]. However, 2D b-P is unstable in air and reacts with oxygen. The researchers

$ \begin{array}{c} \operatorname{ex} \operatorname{b-A}_{0,33} \operatorname{P}_{0,17} & 5-20 & \operatorname{Ti-Au} & \operatorname{BD-SiO} & \operatorname{PMMA} & h = 307 & [13] \\ \operatorname{ex} \operatorname{b-A}_{0,33} \operatorname{P}_{0,17} & 37 & \operatorname{Cr-Au} & \operatorname{hBN-SiO} & \operatorname{hBN} & h = 79, e = 83 & [29] \\ \operatorname{ex} \operatorname{b-A}_{0,33} \operatorname{P}_{0,17} & 15 & \operatorname{Ti-Au} & \operatorname{SiO} & \operatorname{PMMA} & h = 110 & [12] \\ \operatorname{cal} \operatorname{b-ASP} & 1-\operatorname{layer} & - & - & - & h = 2100, e = 14380 & [59] \\ \operatorname{ex} \operatorname{b-P} & 5-25 & \operatorname{Ti-Au} & \operatorname{Al}_2O_3 & \operatorname{Al}_2O_3 & h = 200 & [60] \\ \operatorname{ex} \operatorname{b-P} & 8 & \operatorname{Ti-Au} & \operatorname{SiO} & - & h = 0, e = 0.5 & [61] \\ \operatorname{ex} \operatorname{b-P} & 4.5 & \operatorname{Ti-Au} & \operatorname{SiO} & - & h = 100, e = 0.5 & [61] \\ \operatorname{ex} \operatorname{b-P} & 6-7 & \operatorname{Ti-Au} & \operatorname{SiO} & - & h = 25, e = 0.12 & [63] \\ \operatorname{ex} \operatorname{b-P} & 20 & \operatorname{Ti-Au} & \operatorname{SiO} & - & h = 25, e = 0.12 & [63] \\ \operatorname{ex} \operatorname{b-P} & 20 & \operatorname{Ti-Au} & \operatorname{SiO} & - & h = 52 & [44] \\ \operatorname{ex} \operatorname{b-P} & 20 & \operatorname{Ti-Au} & \operatorname{SiO} & - & h = 210, e = 1 & \\ \operatorname{ex} \operatorname{b-P} & 4.8 & \operatorname{Ti-Au} & \operatorname{SiO} & \operatorname{Cs}_2\operatorname{CO} & e = 27 & [65] \\ \hline \end{array} \\ \begin{array}{c} \operatorname{ex} \operatorname{b-P} & 4.8 & \operatorname{Ti-Au} & \operatorname{SiO} & - & h = 561 & [66] \\ \operatorname{ex} \operatorname{b-P} & - & \operatorname{Au} & \operatorname{SiO} & \operatorname{Al}_2O_3 & h = 100 & [67] \\ \operatorname{ex} \operatorname{b-P} & 1.3 & \operatorname{Ti-Au} & \operatorname{Al}_2O_3 & \operatorname{Al}_2O_3 & h = 100 & [67] \\ \operatorname{ex} \operatorname{b-P} & 10 & \operatorname{Ti-Au} & \operatorname{SiO} & \operatorname{Al}_2O_3 & h = 100 & [67] \\ \operatorname{ex} \operatorname{b-P} & 7 & \operatorname{Au} & \operatorname{SiO} & \operatorname{Al}_2O_3 & h = 230 & [56] \\ \operatorname{ex} \operatorname{b-P} & 7 & \operatorname{Au} & \operatorname{SiO} & \operatorname{Al}_2O_3 & h = 286 & [9] \\ \operatorname{ex} \operatorname{b-P} & 5 & \operatorname{Ti-Pd} & \operatorname{SiO} & - & h = 286 & [9] \\ \operatorname{ex} \operatorname{b-P} & 5 & \operatorname{Ti-Pd} & \operatorname{SiO} & - & h = 286 & [9] \\ \operatorname{ex} \operatorname{b-P} & 5 & \operatorname{Ti-Pd} & \operatorname{SiO} & - & h = 200 & [68] \\ \operatorname{ex} \operatorname{b-P} & 3 & \operatorname{Cr-Au} & \operatorname{hBN-SiO} & \operatorname{hBN} & h = 1350 & [68] \\ \operatorname{ex} \operatorname{b-P} & 3 & \operatorname{Cr-Au} & \operatorname{hBN-SiO} & \operatorname{hBN} & h = 1350 & [68] \\ \operatorname{ex} \operatorname{b-P} & 5 & \operatorname{Ti-Pd} & \operatorname{SiO} & - & h = 260 & [71] \\ \operatorname{ex} \operatorname{Al} \operatorname{doped} \operatorname{b-P} & - & \operatorname{Au} \operatorname{hBN-SiO} & \operatorname{hBN} & h = 1350 & [68] \\ \operatorname{ex} \operatorname{b-P} & 3 & \operatorname{AL} & \operatorname{SiO} & \operatorname{Al}_2O_3 & e = 1495 & [72] \\ \operatorname{ex} \operatorname{Cu} \operatorname{doped} \operatorname{b-P} & 10 & \operatorname{Ti-Au} & \operatorname{hBN} & \operatorname{SiO} & - & h = 400, e = 83 & [46] \\ \operatorname{ex} \operatorname{b-P} & \operatorname{Al} & \operatorname{hBN} & - & e = 690 & [73] \\ \operatorname{ex} \operatorname{b-P} & \operatorname{Al} & \operatorname{Al} & \operatorname{Al} O_2 & \operatorname{Al}_2O_3 & e$	Materials	Thickness $(nm)$	Contact electrode	Dielectric layer	Passivation layer	Mobility $(cm^2 \cdot V^{-1} \cdot s^{-1})$	Ref.	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-As_{0.83}P_{0.17}	5 - 20	Ti-Au	$SiO_2$	PMMA	h = 307	[13]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-As_{0.83} P_{0.17}	37	Cr-Au	$hBN-SiO_2$	hBN	h = 79, e = 83	[29]	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	ex b-As_{0.83}P_{0.17}	15	Ti-Au	$SiO_2$	PMMA	h = 110	[12]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	cal b-AsP	1-layer	_	_	_	h = 2100, e = 14380	[59]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	5 - 25	Ti-Au	$Al_2O_3$	$Al_2O_3$	h = 200	[60]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	8	Ti-Au	$SiO_2$	_	h = 100, e = 0.5	[61]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	4.5	Ti-Au	$SiO_2$	PMMA	h = 142	[62]	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	ex b-P	6-7	Ti-Au	hBN	_	h = 25, e = 0.12	[63]	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	ex b-P	5 - 15	Ti-Au	$SiO_2$	_	h = 52	[44]	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	ex b-P	20	Ti-Au	$SiO_2$	$Al_2O_3$	h = 0.96	[64]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			Ti-Au	$\mathrm{SiO}_2$	_	h = 215, e = 1		
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	ex b-P	4.8			$Cs_2CO_3$	e = 27	[65]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$					$MoO_3$	h = 200		
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex Se doped b-P	_	Cr-Au	$SiO_2$	_	h = 561	[66]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	_	Au	$SiO_2$	$Al_2O_3$	h = 100	[67]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	11.3	Ti-Au	$Al_2O_3$	$Al_2O_3$	h = 187	[53]	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	7	Au	$SiO_2$	$Al_2O_3$	h = 230	[56]	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	10	Ti-Au	$SiO_2$	_	h = 286	[9]	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	6.5	Cr-Au	$SiO_2$	_	h = 984	[10]	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	5	Ti-Pd	$SiO_2$	_	h = 205	[19]	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	1.9	Ti	$SiO_2$	$Al_2O_3$	h = 172, e = 38	[42]	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	8	Cr-Au	$hBN-SiO_2$	hBN	h = 1350	[68]	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	_	Au	$hBN-SiO_2$	_	h = 400, e = 83	[46]	
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	ex b-P	$43 \pm 2$	Ti-Au	$SiO_2$	MMA-PMMA	h = 900	[69]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex b-P	_	Cr-Au	$hBN-SiO_2$	hBN	h = 5200	[70]	
3 $H_2^{10}$ $e = 275$ $[11]$ ex Al doped b-P       5       Ti-Au       SiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> $e = 1495$ $[72]$ ex Cu doped b-P       10       Ti-Au       hBN       - $e = 690$ $[73]$ ex b-P       18.7       Ni       SiO <sub>2</sub> - $h = 170$ $[74]$ Pd       -       Pd       - $h = 186$ [74]         ex b-P       15       Ti-Au       Al <sub>2</sub> O <sub>3</sub> Al <sub>2</sub> O <sub>3</sub> $h = 310, e = 89$ [50]         ex b-P       13       Ti-Au       Al <sub>2</sub> O <sub>3</sub> PMMA $h = 233$ [52]	ex b-P	13 3	AL	$\mathrm{SiO}_2$	$Al_2O_3$	e = 950	[71]	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$						e = 275	[• +]	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex Al doped b-P	5	Ti-Au	$SiO_2$	$Al_2O_3$	e = 1495	[72]	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	ex Cu doped b-P	10	Ti-Au	hBN	_	e = 690	[73]	
Pd $h = 186$ ex b-P     15     Ti-Au     Al <sub>2</sub> O <sub>3</sub> Al <sub>2</sub> O <sub>3</sub> $h = 310, e = 89$ [50]       ex b-P     13     Ti-Au     Al <sub>2</sub> O <sub>3</sub> PMMA $h = 233$ [52]	ex b-P	18.7	Ni	SiOa	_	h = 170	[74]	
ex b-P       15       Ti-Au       Al <sub>2</sub> O <sub>3</sub> Al <sub>2</sub> O <sub>3</sub> $h = 310, e = 89$ [50]         ex b-P       13       Ti-Au       Al <sub>2</sub> O <sub>3</sub> PMMA $h = 233$ [52]			Pd	5102		h = 186		
ex b-P 13 Ti-Au $Al_2O_3$ PMMA $h = 233$ [52]	ex b-P	15	Ti-Au	Al <sub>2</sub> O <sub>3</sub>	Al <sub>2</sub> O <sub>3</sub>	h = 310, e = 89	[50]	
	ex b-P	13	Ti-Au	$Al_2O_3$	PMMA	h = 233	[52]	
cal b-P I-tayer $    h = 26000, e = 1140$ [58]	cal b-P	1-layer		_	_	h = 26000, e = 1140	[58]	
multi-layer $h = 6400, e = 1580$		multi-layer				h = 6400, e = 1580	[00]	

 ${\bf Table \ 1} \quad {\rm Electrical \ performance \ of \ b-AsP \ and \ b-P \ based \ field-effect \ transistors^{a)}}$ 

a) ex: experimental results; cal: calculated results; h: hole mobility, e: electron mobility; PMMA: polymethyl methacrylate; MMA: methyl methacrylate.

found that there is a chemisorption process between oxygen and the lone pairs of phosphorus atoms at the surface of b-P [78–81]. The chemisorption of oxygen leads to the formation of oxygen defects, which makes the surface of phosphorene hydrophilic and forms a mixture of oxide and phosphoric acid. At the same time, the oxidation of 2D b-P can be accelerated by photons, and the phosphorus oxide can further be transformed into aqueous phosphoric acid with the help of water vapor, resulting in the poor stability of naked b-P devices in air [82]. Although the detailed degradation process of b-AsP devices has not been reported, their behavior may be similar to that of b-P devices. Up to now, several surface passivation strategies for b-P and b-AsP have been reported and the stability of devices can be dramatically improved. It has been shown that b-P encapsulated by aluminum oxide  $(Al_2O_3)$  keeps long term stable in air [29, 50, 67, 83, 84]. Without encapsulation, bubbles develop quickly on the b-P surface [60]. Moreover, b-P and b-AsP sandwiched between hBN layers exhibits long term stability in ambient conditions and even under the irradiation of strong laser [45,68,85,86]. In addition to covering b-P with hBN or oxide by atomic layer deposition, organic molecules also are a good choice to prevent degradation and keep long term device stability [13, 16, 55, 69, 87, 88]. There are many encapsulation methods that have demonstrated long term effectiveness and can be used to device development in the future. Moreover, different types of defects in b-P can affect the stability in theory [89, 90]. Therefore,

Material	Th (nm)	Device structure	Wavelength (nm)	R (mA/W)	EQE (%)	Speed (ms)	$D^* (\mathrm{cm}^2 \mathrm{V}^{-1} \mathrm{s}^{-1})$	Ref.
b-AsP	25 - 35	Phototransistor	4600	_	8	0.0124,  0.0089	$2.4 \times 10^{10}$	[27]
b-AsP	188	Phototransistor	3400	190		_		
			5000	16	_		_	[29]
			7700	1.2				
b-AsP	5 - 20	Phototransistor	3662	180	6.1	0.54,  0.52	$1.06 \times 10^{8}$	[13]
$MoS_2$ -b- $AsP$	_	Photodiode	4290	115.4	3.33	_	$4.9 \times 10^{9}$	[13]
			2360	216.1	11.36	_	$9.2 \times 10^{9}$	
$MoS_2$ -b- $AsP$	66, 59	Photodiode	520	0.3	71	$0.009, \ 0.005$	_	[16]
InSe-b-AsP	10, 11.5	Photodiode	520	1000	1.5	$0.217, \ 0.089$	$1 \times 10^{12}$	[14]

Table 2 Performance of b-AsP based photodetectors<sup>a)</sup>

a) Th: thickness. R: responsivity,  $R = I_{\rm ph}/P_{\rm in}$ , where  $I_{\rm ph}$  is photocurrent and  $P_{\rm in}$  is incident light power. EQE =  $hcR\lambda^{-1}e^{-1}$ , where h is the Planck constant, c is the speed of light, and e is the electron charge.  $D^*$ : detectivity, defined by  $D^* = \frac{R\sqrt{AB}}{S_n}$ , where A is the active area, B is the electrical bandwidth, and  $S_n$  is the noise spectral density. Response speed: rise time is defined as from 10% to 90% of the photocurrent and fall time is defined as from 90% to 10% of the photocurrent.

defects in b-P plays a key role in the degradation process, which indicates that the stability of b-P can be enhanced by improving the material quality.

# **3** Photodetector applications

Photodetectors are widely used in applications of imaging, optical guidance, optical communication, remote sensing, etc. Various 2D materials such as graphene [91,92], transition metal dichalcogenides [93–95], b-P and b-AsP have been exploited for photodetectors. High responsivity, fast response speed and broadband of detection have been the main challenge for 2D materials based photodetectors. The disadvantage of graphene is that it has no natural bandgap with a low response to light, while transition metal dichalcogenides are limited by large bandgaps so that they are mainly used for visible light detection. In contrast, b-P possesses a tunable bandgap at about 0.3–2.0 eV, allowing for photo detection in a large spectral range from ultraviolet to infrared [20, 38, 43, 44, 56, 61, 62, 67, 84, 92–111]. Interestingly, the b-AsP's bandgap is about 0.15–0.3 eV, which is smaller than that of b-P and can be used for mid-infrared photodetectors [12, 13, 27].

For b-P based photodetectors, various studies on phototransistors, photodiodes, plasmonics [40, 64, 101], and waveguides [104, 109–111] have been reported. However, less studies are reported on b-AsP based photodetectors, which mainly focus on phototransistors and photodiodes. In this section, we will introduce phototransistor and photodiode applications of b-P and b-AsP. We compared the structure and performance of photodetectors of these two materials to find new research directions for b-AsP based photodetectors.

The performance of photodetectors is commonly characterized by several figures of merit, such as external quantum efficiency (EQE), responsivity (R), response speed and detectivity  $(D^*)$ . In Table 2 [13, 14, 16, 27, 29], their definitions and computing methods are summarized.

### 3.1 Phototransistors

When a photoconductor is exposed to light, the photo excited electron and hole can increase its conductivity. Under an external bias, the photo exited electron and hole will draft to opposite contact electrodes, and result in the generation of photocurrent. If one type of photo generated carries is trapped and another one could circulate many times through an external circuit before recombination with trapped ones, the device can produce a photo gain.

To improve the photosensitivity of photodetectors, two strategies are general used: one is maximizing their electrical response to light by improving the photo gain effect, another one is to minimize the noise of electrical output. The carrier density in the semiconductor can be modulated effectively by applying a gate voltage  $V_G$ , so that the device can be worked in the depletion regime by switching off the dark current. Therefore, the photo gain effect and low noise current can be achieved in phototransistors simultaneously, resulting in a high photosensitivity.

For b-P based phototransistors, a large responsivity of  $9 \times 10^4$  A/W in the ultraviolet region has been reported (Figure 5(a-i)) [62]. With increasing incident light wavelength, the photoresponse of the photo-



Figure 5 (Color online) Phototransistors based on b-P and b-AsP. (a-i) Device structure of few-layer b-P based phototransistor operating at UV light. (a-ii) Device structure of the b-P based photodetector for infrared detection. Inset: optical image of the phototransistor. (a-iii) The responsivity as a function of incident light power at  $V_{ds} = 100 \text{ mV}$  and  $V_{ds} = 500 \text{ mV}$ , respectively. (b-i) Photo response of a b-AsP based phototransistor operating at 8.05 µm with a power density of 0.17 W·cm<sup>-2</sup>. Inset: optical image of this device. (b-ii) Cross-sectional diagram of the phototransistor based on hBN/b-As<sub>0.83</sub>P<sub>0.17</sub>/hBN heterostructure. (b-iii) The photocurrent as a function of  $V_{gs}$ . (a-i) is reproduced with permission from [62], (a-ii) and (a-iii) from [57], (b-i) from [13], and (b-iii) from [29].

transistor decreases rapidly, but the response rate is improved. Buscema et al. [61] reported a b-P based phototransistor with ultrafast photoresponse. A rise time of 1 ms and fall time of 4 ms were achieved at a wavelength of 940 nm, while the responsivity of the phototransistor is only 4.8 mA/W. The tradeoff between responsivity and response speed is very common in phototransistors, and this relationship is usually attributed to the photogating mechanism in 2D phototransistors. Guo et al. [57] first fabricated a b-P based mid-infrared phototransistor with a 10 nm-thick nanosheet (Figure 5(a-ii)). The device can detect an infrared light at a wavelength of 3.39  $\mu$ m. At the same time, this device exhibits high photo gain with a responsivity of 82 A/W (Figure 5(a-iii)) [57]. Besides the high photosensitivity, the device can be effectively modulated at a frequency of kilohertz, owing to the fast carrier dynamics of b-P. The detection range of mid infrared spectrum is technological interest and scientific importance, which involves many important applications, including optical radar, free space communication, remote sensing and molecular fingerprint imaging.

Compared with phototransistors based on b-P, a distinguishing feature of b-AsP based phototransistors is their photoresponse at longer wavelengths. The optical image of a typical b-AsP based phototransistor is shown in Figure 5(b-i) (bottom inset). A significant photo response under a mid-infrared light at the wavelength of 8.05 µm is shown in Figure 5(b-i) (upper inset). Furthermore, the phototransistor shows a high response speed with a rise time of 0.54 ms and a delay time of 0.52 ms. As the increasing of bias voltage, the photocurrent  $(I_{\rm P} = I_{\rm light} - I_{\rm dark})$  increases linearly at  $V_{\rm gs} = 0$  V (Figure 5(b-i)) [13]. Yuan et al. [29] have fabricated a b-As<sub>0.83</sub> $P_{0.17}$  photodetector sandwiched between layers of hBN (Figure 5(bii)). With the increasing of wavelength of incident light from 3.4 to 7.7 µm, the peak responsivity of phototransistor decreases from 190 to 1.2 mA/W at room temperature (Figure 5(b-iii)). Owing to the protection with hBN, the  $b-As_{0.83}P_{0.17}$  photodetector reaches excellent air stability. In addition, the photo response of b-AsP can be maximized by adjusting the thickness of the material, and the dark current can also be minimized leading to a large detectivity. Amani et al. [27] reported a peak detectivity of  $2.4 \times 10^{10} \text{ cm} \cdot \text{Hz}^{1/2} \cdot \text{W}^{-1}$  for phototransistors of 36 nm-thick b-AsP (91% As) at room temperature. Compared with commercial mid-infrared detectors, the peak detectivity of the b-AsP (91% As) based phototransistor is improved by one order of magnitude [27]. In summary, the broad photoresponse in the mid-infrared region arising from the intrinsic bandgap, together with the improvement of stability in air condition, makes b-AsP a promising material for photodetector application in the mid-infrared range.

## 3.2 Photodiodes

In photodiodes, space charge region is formed at the p-n junction or Schottky junction between metal and semiconductors, resulting in build-in field. Under the light, photo generated carriers are driven by the built-in potential, moving to opposite electrodes.

Figure 6(a-i) shows an electrically tunable diode based on a vdW p-n heterojunction, fabricating by a p-type b-P and a n-type monolayer  $MoS_2$  flake. The back gate voltage can modulate the currentrectifying characteristics of the device [96]. With the decreasing of back gate voltage, the rectification ratio (the ratio of the forward to reverse current) increases obviously. In the device, a peak rectification ratio of  $10^5$  can be obtained at  $V_{\rm gs} = -30$  V through applying a bias voltage from -2 V to 2 V. Under a 633 nm laser with different incident power, the current of the photodiode increases obviously as shown in Figure 6(a-ii). A peak responsivity of 418 mA/W and external quantum efficiency of 0.3% at a bias of -2 V, can be abstracted from the *I-V* curves of the p-n diode [96]. Furthermore, a few-layer MoS<sub>2</sub>/b-P heterojunction based photodiode is fabricated to detect near-infrared at room temperature, achieving a peak detectivity of  $2.13 \times 10^9$  cm·Hz<sup>1/2</sup>·W<sup>-1</sup> at  $\lambda = 1550$  nm [99]. By using a b-P vertical p-n junction, Yuan et al. [97] designed a polarization-sensitive broadband photodetector. In order to collect the photo generated hot carrier in an isotropic manner, the metal electrodes acting as the photocurrent collector are designed as ring shape; therefore the possibility of the photocurrent arising from geometric edge effect at the metal-b-P edge can be excluded [97]. Within the ring shape electrode, the photocurrent excited by  $90^{\circ}$  polarization light is far less than that by  $0^{\circ}$  light polarization, indicating that the intrinsic polarization dependent photo response arises from the b-P itself [97]. Photodetectors based on in-plane b-P p-n homojunctions have also been reported, which are defined by local electrostatic gating or chemical doping methods and exhibit an enhanced photo response [63, 105]. In addition, high-quality InSe/b-P vertical PN heterojunction has been reported. The heterojunction has an atomic-level flat interface and a perfect crystal lattice. The ballistic avalanche phenomenon is observed, and five orders of magnitude current jumps occur in this device. This device can be used as mid infrared photodetector, showing very high photon amplification (more than 10000) and lower noise performance than the theoretical limit of traditional avalanche photodetectors [57].

As for b-AsP-based vdW photodetectors, an attractive research direction is to build mid-infrared photodiodes. Long et al. [13] fabricated photodiodes using p-type b-AsP and n-type MoS<sub>2</sub>. A vdW p-n junction can be formed using b-AsP and MoS<sub>2</sub>, and the current under forward bias is higher than that under reverse bias more than two orders of magnitude. In the 3 to 5  $\mu$ m range, the detectivity of this device is consistently larger than 4.9  $\times$  10<sup>9</sup> cm·Hz<sup>1/2</sup>·W<sup>-1</sup> with a peak value about 9.2  $\times$  10<sup>9</sup> cm·Hz<sup>1/2</sup>·W<sup>-1</sup> (Figure 6(b-i)) [13].

With the proper band alignment design of b-AsP-based heterojunctions, high-performance photodiodes can be realized. Wu et al. [14] designed a vertically stacked vdW heterojunction tunneling device of b-AsP and indium selenide (Figure 6(b-ii), inset), showing an unusual ultralow forward current below picoampere. Moreover, a record high reverse rectification ratio more than  $10^7$  is achieved in this device (Figure 6(b-ii)). The vdW heterojunction tunneling device functions as a sensitive photodiode in the visible wavelength range, as a result, an ultrahigh light on/off ratio about  $1 \times 10^7$  is achieved. At the same time, the device exhibits a high detectivity of over  $1 \times 10^{12}$  cm·Hz<sup>1/2</sup>·W<sup>-1</sup> and a comparable responsivity of about 1 A/W [14]. Wu et al. [16] also designed a sensitive photodiode based on b-AsP/  $MoS_2$  vdW heterojunction, in which the narrow gap b-AsP was utilized as an effective carrier selective contact with a unilateral depletion region band design. From the atomic force microscope image of this device, the thicknesses of b-AsP and  $MoS_2$  are measured to be 59 and 66 nm, respectively. The heterojunction shows high sensitive at visible wavelength, but is not sensitive at infrared wavelength, because the photo response of this device mainly comes from  $MoS_2$  flake [16]. To reveal the mechanism of this device, energy band diagram of  $b-AsP/MoS_2$  heterojunction is drawn. When the device is illuminated under a 520-nm light, the photo carriers in the device mainly are generated in the thick  $MoS_2$  channel. Because a large unilateral depletion region is formed in the  $MoS_2$  side, the photo generated electrons in MoS<sub>2</sub> can draft to the b-AsP side freely and be collected by the drain electrode, while the photo generated holes will draft from the  $MoS_2$  side to the source electrode [16]. This device achieves a record high response speed for photovoltaic detectors based on 2D vdW heterostructures, with a rise time of 9  $\mu$ s and a fall time of 5  $\mu$ s. This ultrafast response speed results from that the interface trapping effect is reduced significantly in the unilateral depletion region, and the photo generated electrons could recombine with the holes in the accumulation region rapidly [16].



Figure 6 (Color online) Photodiodes based on b-P and b-AsP. (a-i) Schematic of the monolayer MoS<sub>2</sub>/b-AsP photodiode. (a-ii)  $I_{ds}$ - $V_{ds}$  curves of the p-n photodiode based on monolayer MoS<sub>2</sub>/b-AsP under the incident light with various powers. Inset: the detailed reverse region at bias from -1 V to 0 V. (b-i) The detectivity as a function of wavelength at  $V_{ds} = 0$  V. Compared with the detectivity of commercial thermistor bolometer [106] and PbSe mid-infrared detectors, the MoS<sub>2</sub>/b-AsP photodiode shows great advantages. (b-ii)  $I_{ds}$ - $V_{ds}$  curve of the InSe/b-AsP diode at  $V_{gs} = 10$  V. Inset: schematic of the device. (a-i) and (a-ii) are reproduced with permission from [96], (b-i) from [13], and (b-iii) from [14].

# 4 Perspectives

First, the highest reported mobility of b-P is about  $5200 \text{ cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1}$ , which approaches its theoretical value. However, the highest mobility reported for b-AsP to date is only  $307 \text{ cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1}$ , which is much lower than the theoretically calculated value of  $10000 \text{ cm}^2 \cdot \text{V}^{-1} \cdot \text{s}^{-1}$ . As shown in Table 1, the mobility of b-P can be improved through proper contact electrodes, passivation layers and doping methods, which provides some directions for the improvement of the electronic properties of b-AsP. In addition, electronic anisotropy is another interesting property of b-AsP. The lattice periodicities are different along with different crystallographic directions, which results in anisotropy in crystal lattice. Generally, anisotropy in 2D materials is more pronounced owing to their atomic thickness. Electronic anisotropy is important for future multifunctional directional photodetectors based on b-AsP. Beyond that, according to the existing work in theory, there are still some properties and models to be verified for b-AsP, including the optical and transport bandgaps of monolayer material, the crystal structures with different components, and the key factors limiting the hole and electron mobility.

Second, as shown in Table 2, b-AsP based mid-infrared photodetectors show great advantages with respect to other 2D materials because of its small bandgap. However, detectivity and responsivity of b-AsP based mid-infrared photodetectors are lower than  $1 \times 10^{10}$  cm·Hz<sup>1/2</sup>·W<sup>-1</sup> and 1 A/W, respectively, indicating a weak photo detection ability. To improve the photo detection ability of b-AsP, enhancing the photogating effect of the photodetectors presents a valuable research direction. Several methods such as designing waveguide or nanoplasmonics, p-type or n-type doping, or metal-doping, have been applied to enhance the photodetector performance of b-P (Table 3 [54,56,61–66,84,96–98,101,102,104,105,107–109,111]) and may also be applicable to b-AsP. To achieve high-performance infrared photodetectors, the thickness of b-AsP needs to be optimized, because it influences the optoelectronic properties of the photodetectors (Table 2). In addition, flexible mid-infrared photodetectors are one of the most important research directions in the field of wearable devices in the future, and they have great potential in the human health sector for pulse detection, blood oxygen detection, and body temperature detection.

Material	Th (nm)	Device structure	Wavelength (nm)	R (mA/W)	EQE (%)	Speed (ms)	$D^* (\mathrm{cm}^2 \mathrm{V}^{-1} \mathrm{s}^{-1})$	Ref.
b-P	8	Phototransistor	640	4.8	_	1, 4	_	[61]
b-P	4.5	Phototransistor	UV	$9 \times 10^7$	_	1, 4	$3 \times 10^{13}$	[62]
p-n b-P	10	Phototransistor	1550	$5 \times 10^3$	390	0.035,  0.04	_	[105]
b-P	12	Phototransistor	3390	$8.2 \times 10^4$	_	_	_	[107]
$b-P-C_{S2}CO_3$	4.0	Phototransistor	405	1.88	576			[65]
$b$ - $P$ - $MoO_3$	4.0			2.56	784	_		
Se doped b-P	_	Phototransistor	635	$1.53 \times 10^{4}$	2993	_	_	[66]
b-P	_	Phototransistor	1550	6	_	0.1,  0.3	_	[84]
WCo DD Mog	_	Phototransistor	532	$6.32 \times 10^{3}$		_	$1.25 \times 10^{11}$	[102]
$WSe_2$ -BP-MoS <sub>2</sub>			1550	$1.12 \times 10^3$	_	_	$2.1 \times 10^{10}$	
b-P	10	Phototransistor	830	$5.3 \times 10^4$	_	_	-	[54]
b-P	7	Phototransistor	830	18.7	_	_	_	[56]
b-P	<10	Photodiode	650	3.5	_	2	_	[104]
h D MoS	22, 12	Photodiode	1550	153.4	20	0.015	$3.1 \times 10^{11}$	[98]
D-P-10052			532	$2.23 \times 10^4$	1000	-	$2.13 \times 10^{9}$	
p-n b-P	_	Photodiode	940	-	0.1	-	_	[63]
b-P-MoS <sub>2</sub>	11, 0.9	Photodiode	633	418	0.3	-	_	[96]
p-n b-P	30	Photodiode	1200	0.35	_	-	_	[97]
b-P-InSe	_	Photodiode	455	11.7	3.2	24, 32	_	[108]
b-P	135	Plasmonics	1550	12	_	-	_	[101]
b-P	20	Plasmonics	1550	$1 \times 10^4$	_	_	_	[64]
b-P	40	Waveguide	3825	$1.13 \times 10^4$	_	_	_	[109]
b-P	36.8	Waveguide	2000	306.7	_	_	_	[111]
b-P	23	Wayeguide	3700	$2 \times 10^3$	_		_	[107]
	40	waveguide	3700	$2.3 \times 10^4$				
b-P	11	Wayoguida	1570 - 1580	135	_	-	_	[104]
	100	waveguide		675	_	$3 \times 10^{-7}$	_	[104]

Table 3 Performance of b-P based photodetectors<sup>a)</sup>

a) p-n: p-n junction. Th: thickness. R: responsivity. EQE: external quantum efficiency.  $D^*$ : detectivity.

Flexible photodetectors based on b-AsP have not been studied to date and require further investigation.

Third, the distinctive band structure of b-AsP provides many opportunities for further band structure engineering. In combination with other 2D materials, the 2D heterostructure devices can be designed. 2D heterojunctions, such as b-AsP/MoS<sub>2</sub> or b-AsP/InSe, have been fabricated with excellent rectification effect and photodetector performance. Besides 2D structures, 3D/b-AsP, 1D/b-AsP, and even 0D/b-AsP heterostructures are future research focuses for the development of novel devices beyond photodetectors.

Finally, in addition to fabricating novel devices and improving the performance of existing devices, there are still challenges to realize their practical applications. One of the major challenges is that the few-layer b-P and b-AsP are unstable in atmosphere. The performance of devices will be degraded because of the undesired reaction of b-P (or b-AsP) with water and oxygen. Passivation of the surface of b-P and b-AsP by depositing Al<sub>2</sub>O<sub>3</sub> thin films or encapsulating hBN layer has been proven to be the effective way to protect the material. Another strategy to keep the quality of materials and device performance is to decorate the surface with organic molecules. But, these methods still cannot prevent the degradation of b-P and b-AsP absolutely over a long period of time. It is still necessary to explore more effective and convenient passivation strategies. Another one challenge is to grow large area 2D b-P and b-AsP. Mechanical exfoliation method has been widely used to fabricate 2D b-P and b-AsP samples, however, the samples just can be used to demonstrate proof-of-concept device because of their small size. To realize practical electronic or optoelectronic systems, it is essential to grow large area b-P and b-AsP, which still is a huge challenge. Chemical vapor deposition (CVD) method has been demonstrated to grow high quality 2D materials with large area. Maybe it is a promising direction to fabricate large area 2D b-P and b-AsP in the future.

# 5 Summary

In this review, we summarized the properties of 2D b-P and b-AsP and the recent progress on their applications as photodetectors. The distinguished properties of 2D b-AsP, including high carrier mobility and proper bandgap make it a good tradeoff between transition metal dichalcogenides and graphene, as well as a promising material for high performance electronic and optoelectronic devices. Although some achievements have been made, many challenges and opportunities remain to further improve the photodetector performance.

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